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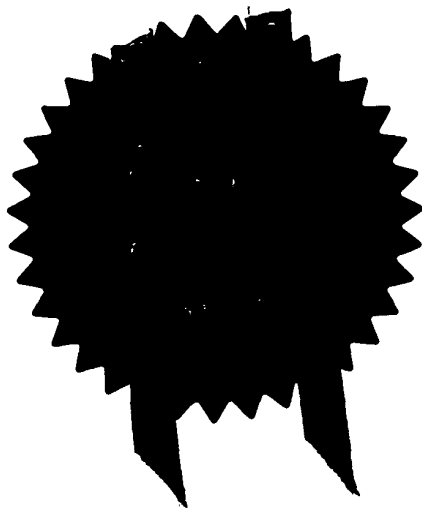
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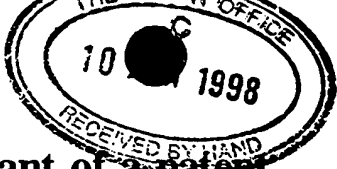


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Request for grant of a patent

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1. Your reference KMN/FP5691530

2. Patent application number **9812523.0**
(The Patent Office will fill in this part) **10 JUN 1998**

3. Full name, address and postcode of the or of each applicant (underline all surnames) ISTITUTO DI RICERCHE DI BIOLOGIA
MOLECOLARE P. ANGELETTI SPA
VIA PONTINA KM. 30.600
00040 POMEZIA (ROMA)
ITALY

Patents ADP number (if you know it) *743921900*
If the applicant is a corporate body, give the country/state of its incorporation ITALY

4. Title of the invention PEPTIDE INHIBITORS OF HEPATITIS C VIRUS NS3 PROTEASE

5. Name of your agent (if you have one) MEWBURN ELLIS
"Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode) YORK HOUSE
23 KINGSWAY
LONDON
WC2B 6HP

Patents ADP number (if you know it) 109006

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7. If this application is divided or otherwise derived from an earlier UK application, give the number and the filing date of the earlier application	Number of earlier application	Date of filing (day / month / year)
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8. Is a statement of inventorship and of right to grant of a patent required in support of this request? (Answer 'Yes' if:

- a) any applicant named in part 3 is not an inventor, or
- b) there is an inventor who is not named as an applicant, or
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Description 41

Claim(s) 0

Abstract 0

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11. I/We request the grant of a patent on the basis of this application.

Signature

Date

Kathryn Nicholls

9 June 1998

12. Name and daytime telephone number of person to contact in the United Kingdom KATHRYN M. NICHOLLS 0117 926 6411

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NS3 PROTEASE

This invention relates to compounds which can act as inhibitors of the hepatitis C virus (HCV) NS3 protease, to uses of such compounds and to their preparation.

The hepatitis C virus (HCV) is the major causative agent of parenterally-transmitted and sporadic non-A, non-B hepatitis (NANB-H). Some 1% of the human population of the planet is believed to be affected. Infection by the virus can result in chronic hepatitis and cirrhosis of the liver, and may lead to hepatocellular carcinoma. Currently no vaccine nor established therapy exists, although partial success has been achieved in a minority of cases by treatment with recombinant interferon- α , either alone or in combination with ribavirin. There is therefore a pressing need for new and broadly-effective therapeutics.

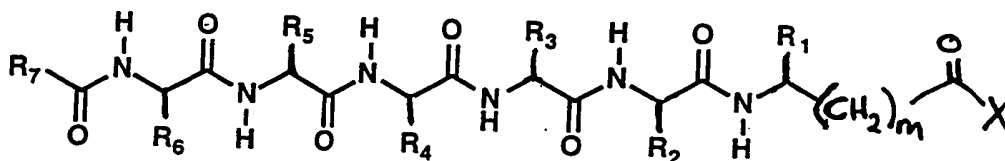
Several virally-encoded enzymes are putative targets for therapeutic intervention, including a metalloprotease (NS2-3), a serine protease (NS3), a helicase (NS3), and an RNA-dependent RNA polymerase (NS5B). The NS3 protease is located in the N-terminal domain of the NS3 protein, and is considered a prime drug target since it is responsible for an intramolecular cleavage at the NS3/4A site and for downstream intermolecular processing at the NS4A/4B, NS4B/5A and NS5A/5B junctions.

Previous research has identified classes of peptides, in

5 particular hexapeptides, showing degrees of activity in inhibiting the NS3 protease. The aim of the present invention is to provide further compounds which exhibit similar, and if possible improved, activity.

10 The present inventors have investigated the replacement of cysteine by 2,2-difluoro-1-aminobutyric acid or 2,2,2-trifluoro-1-aminobutyric acid at the P1 position of certain peptidic product inhibitors and
15 substrates of HCV NS3 serine protease. These studies have shown that fluorocarbon groups, in particular the -CF₂H group may mimic the cysteine thiol group, which is believed to be involved in substrate and inhibitor binding to the S1 specificity pocket of the NS3 protease. In general terms, therefore, the present invention
20 relates to compounds containing fluorocarbon groups, especially -CF₂H and -CF₃, for use as inhibitors of HCV NS3 protease. Examples of such compounds include peptides or peptide analogs, such as hexapeptides, in which a fluorocarbon group, such as -CF₂H, is present as a
25 sidechain, for instance at the C-terminus or P1 position of the peptide.

In particular, according, to a first aspect of the present invention there is provided a compound of formula
30 1:



35 FORMULA 1

the substituents of which are defined below. Optionally, the compound may be in the form of a pharmaceutically acceptable salt. Alternatively it may be in the form of a pharmaceutically acceptable ester.

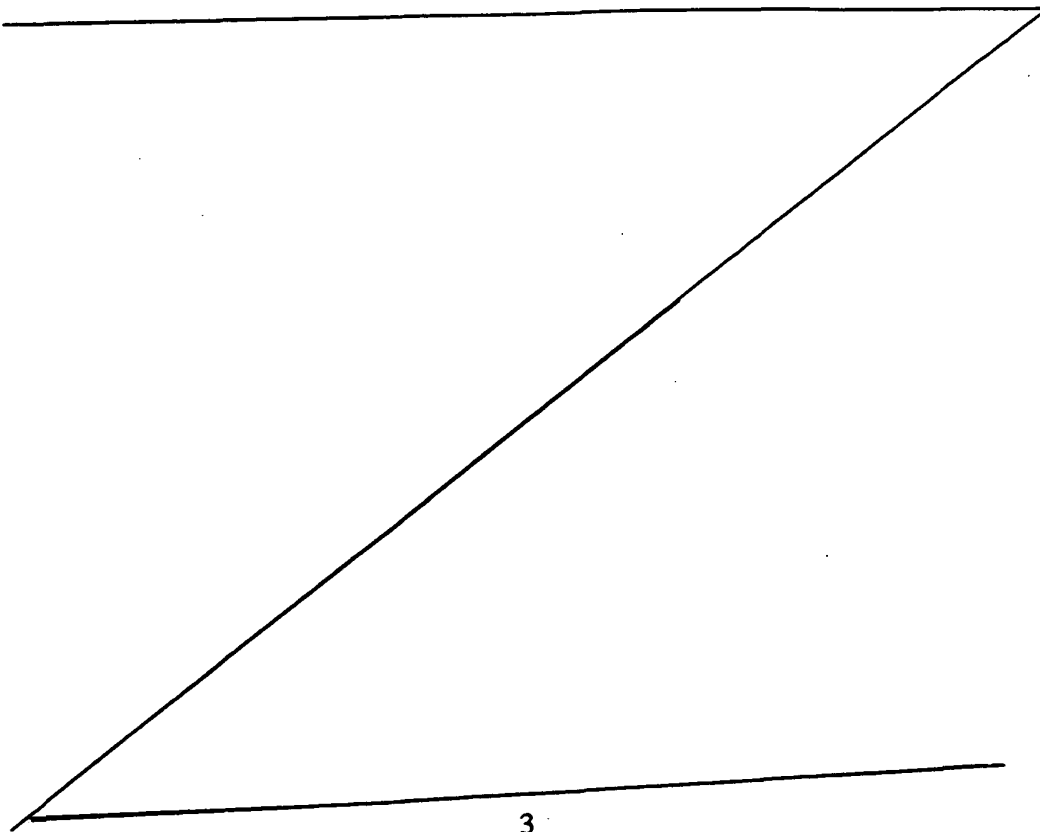
- 10 Formula 1 defines a class of hexapeptides and hexapeptide analogs, examples of which have been found to act as inhibitors of the HCV NS3 protease. The compounds are characterised in particular by the nature of the group R_1 , ie, by the amino acid or amino acid analog found at the
- 15 C- terminus of the hexapeptide.

In formula 1,

m is 0 or 1

20

R_1 is a group containing 1 to 6 carbon atoms and which



5 includes a fluorocarbon group such as $-\text{CF}_3$ or $-\text{CF}_2\text{H}$;

R_2 and R_3 , which may be the same or different, are groups containing hydrocarbon fragments (which may include aromatic groups), and/or acidic functionality;

10

R_4 is a group containing a lipophilic fragment;

R_5 and R_6 may be the same or different and are groups containing an acidic functionality (the acid group may be present in the protonated or unprotonated (ie, carboxylate) form);

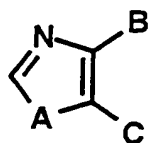
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$\text{R}_7\text{-CON}$ represents an amide, urethane or urea, in which R_7 is a group containing 1 to 8 carbon atoms; and

20

X is OH , CF_3 , H , COOH , CONR_9 , R_{10} , CF_2CONR_9 , R_{10} , or a heterocyclic group, which can be of formula 2:

25



Formula 2

in which A = sulphur, oxygen or NR_8 ,

30

R_8 , R_9 and R_{10} are, independently, hydrogen or any suitable aliphatic or aromatic, optionally substituted, groups, such as in particular alkyl or aralkyl groups, typically having between 1 and 8 carbon atoms, and

B and C are each independently either H or R_{11} , where R_{11} is an alkyl or aralkyl group, typically having between 1

35

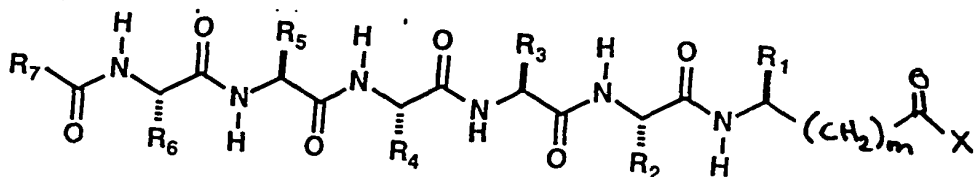
5 and 8 carbon atoms, or can together form part of a ring, preferably an aromatic ring such as a phenyl ring.

The term "acidic functionality" as used herein encompasses not only carboxylic acid groups but also acid
10 mimetics such as tetrazoles and acylsulphonamides.

The groups R_1 - R_6 may be the side chains of naturally-occurring amino acids, or other, preferably analogous, groups having the requisite functionality. Such groups
15 may contain, in addition to carbon and hydrogen, heteroatoms such as nitrogen, oxygen, sulphur and phosphorous; they may be saturated or unsaturated, branched or unbranched, substituted or unsubstituted. Their substituents can include halogens and/or other
20 inorganic elements. Groups of this type preferably contain between 1 and 20 carbon atoms, more preferably between 1 and 13 carbon atoms, particularly preferably between 1 and 8 carbon atoms.

25 Each amino acid, or amino acid analog may have D- or L-stereochemistry, but L-stereochemistry is preferred. Preferably, all five N-terminal amino acids or analogs are L- isomers. The stereochemistry at the R_1 position allows either the D or the L isomer, of which the L
30 isomer is usually preferred. It is particularly preferred that all six amino acids, or analogs are L-isomers. Thus, a preferred stereoisomer of formula 1 is as shown overleaf:

5



10

The compound may be a mixture of stereoisomers,
especially of diastereomers having different
stereochemistry at the position of R_1 substitution only.

Preferred substituents for formula 1 include:

15

$m=0$

for R_1 : $-\text{CH}_2\text{CF}_3$ or, more preferably, $-\text{CH}_2\text{CF}_2\text{H}$;

20

for R_2 : the side chain of an acidic amino acid, especially
- $\text{CH}_2\text{CH}_2\text{CO}_2\text{H}$ (the side chain of glutamic acid), or the
side chain of an amino acid having a non-polar side chain
such as alanine, isoleucine, valine or, in particular, -
 $\text{CH}_2\text{CH}(\text{CH}_3)_2$ (from leucine), or, more preferably, the side
chain of β - cyclohexylalanine:

25



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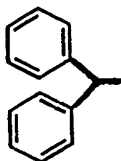
for R_3 : the side chain of an acidic amino acid, for
instance - $\text{CH}_2\text{CH}_2\text{CO}_2\text{H}$ (glutamic acid), or a non-polar
amino acid side chain such as $-\text{CH}(\text{CH}_3)_2$ (from valine), -
 $\text{CH}_2\text{CH}(\text{CH}_3)_2$ (from leucine) or $-\text{CH}(\text{CH}_3)(\text{CH}_2\text{CH}_3)$ (from
isoleucine), preferably glutamic acid;

35

5

for R_4 : the side chain of an amino acid having a hydrophobic R group, for instance - $\text{CH}_2\text{CH}_2\text{SMe}$ (from methionine), or the side chains of leucine, isoleucine, or more preferably diphenylalanine:

10



15

for R_5 and R_6 : - $\text{CH}_2\text{CH}_2\text{CO}_2\text{H}$ (from glutamic acid) or - $\text{CH}_2\text{CO}_2\text{H}$ (from aspartic acid);

⋮

for R_7 : - CH_3 ; and

for R_8 : -H;

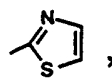
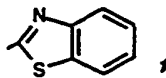
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for R_9 : -H;

for R_{10} : benzyl, phenethyl; and

25

for X: - CO_2H , - CONHCH_2Ph ,



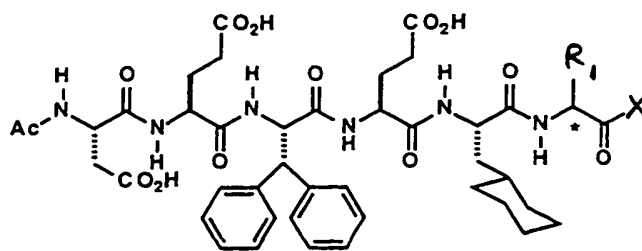
or more preferably H, or -OH.

30

A most preferred group of compounds according to the invention can be represented by the general formula 3:

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5



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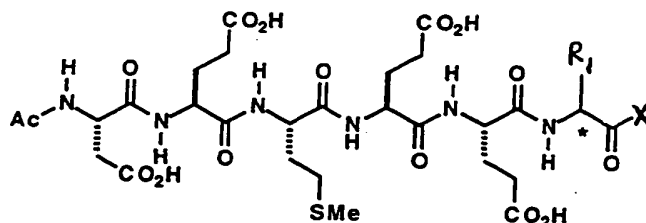
FORMULA 3

in which R_1 and X are as defined above.

Another preferred group of compounds is represented by the general formula 3' in which, again, R_1 and X are as defined above

15

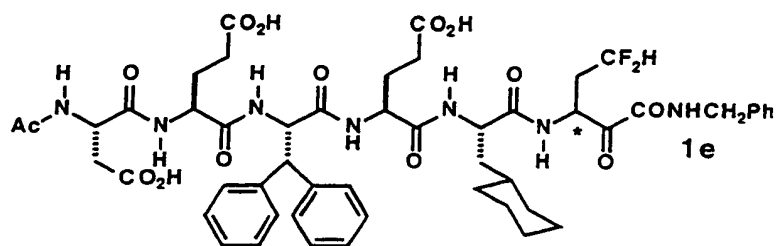
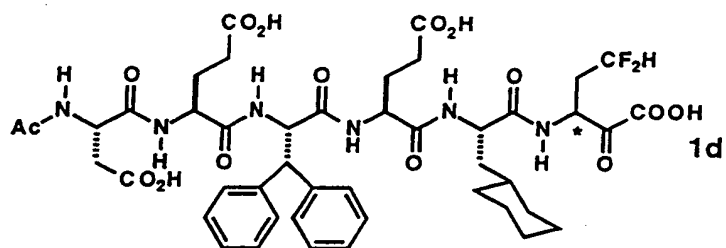
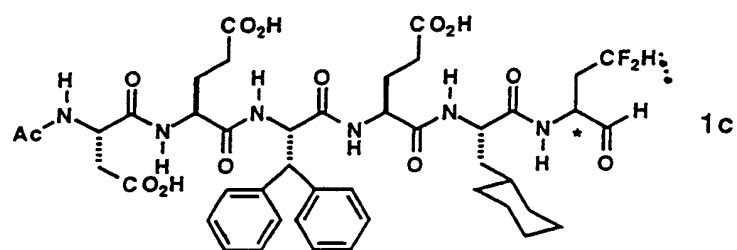
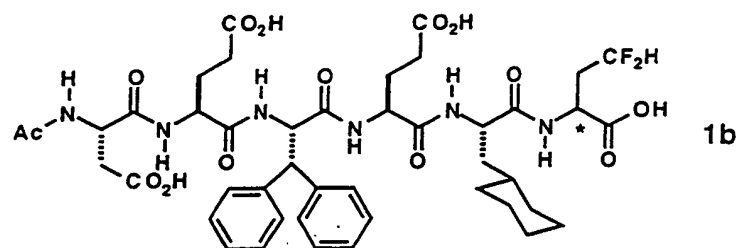
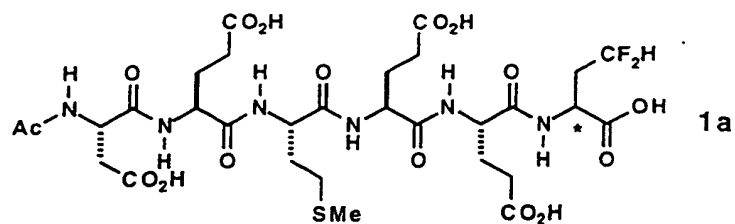
FORMULA 3'



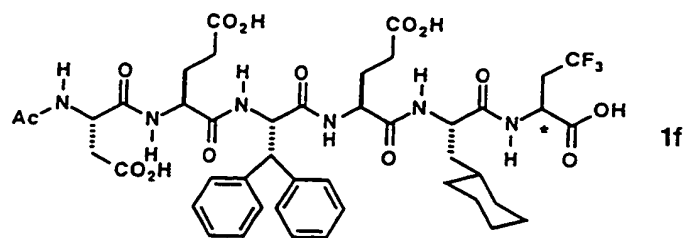
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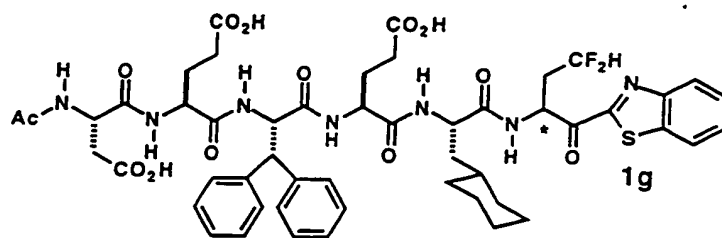
Specific compounds within the scope of the present invention include:



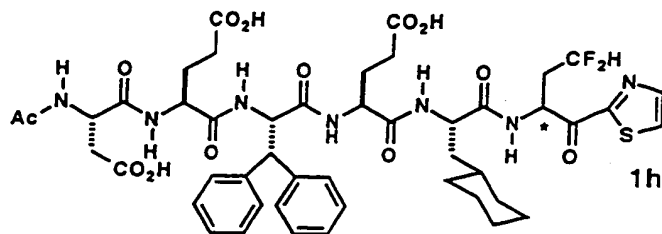
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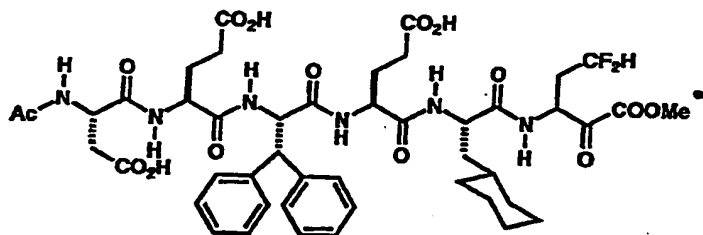
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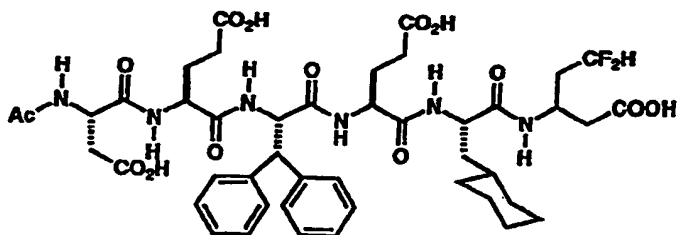


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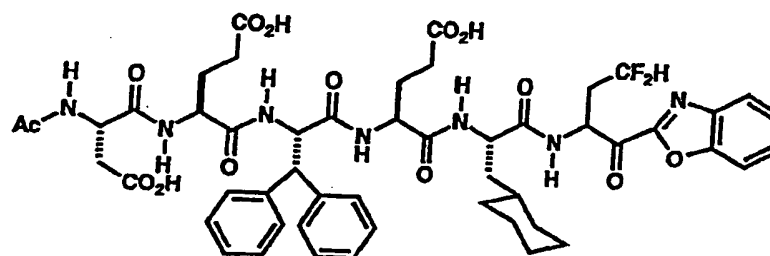


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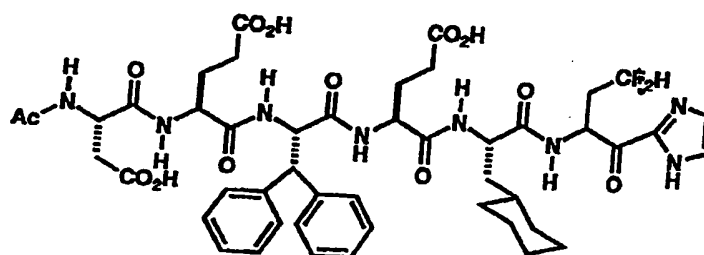
:

10



1l

15



1m

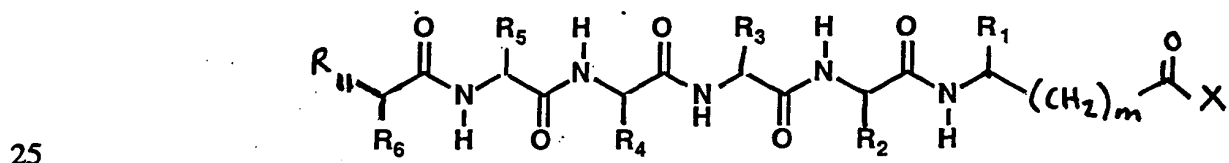
20

These compounds, and analogous compounds within the scope
 of the invention, have been found to inhibit the action
 of the HCV NS3 protease at concentrations lower than
 10 μ M. They can therefore be expected to be of use in
 the treatment and prevention of hepatitis C and other
 related conditions.

The first aspect of the invention also provides a
 derivative of a compound of formula 1. "Derivative"
 includes a compound or composition in which the compound
 of formula 1 is bound to a coupling partner such as a
 label, a supporting substrate, a carrier or an effector
 molecule.

5 In particular, derivatives include "prodrug" forms of the
 compounds of Formula 1 which may be converted in vivo
 into the compound of Formula 1. Examples of such
 derivatives include those in which one or more carboxylic
 acid groups of the compound of Formula 1 are esterified
 10 or otherwise derivatised into groups convertible in vivo
 into carboxylic acid or carboxylate groups. For instance
 carboxylic acid groups may be esterified with C₁-C₁₈
 alcohols, preferably C₁-C₈ alcohols. Another possibility
 is that the derivative may be a C-terminal extended
 15 variant of the compound of Formula 1, convertible in vivo
 into a compound of Formula 1.

Another derivative within the scope of the present
 invention is modified at the N-terminus of the molecule
 20 and has the formula shown below as formula 1'



FORMULA 1'

30 where R₁-R₆, m and X are as specified above, and the same
 qualification concerning stereochemistry apply. However,
 the N-terminal amino acid, providing the group R₆, has
 been replaced by a simple carboxylic acid, with loss of
 the acylamino moiety. In formula 1', R₁ is an optionally
 35 substituted lower alkyl group having from 1 to 6 carbon

5 atoms or, more preferably, is hydrogen. Preferred N-terminal carboxylic acids are glutaric or succinic acid.

According to a second aspect the present invention provides a compound or derivative according to the first aspect, for use in any therapeutic method, preferably for use in inhibiting the HCV NS3 protease, and/or for use in treating or preventing hepatitis C or a related condition. By "related condition" is meant a condition which is or can be caused, directly or indirectly, by the hepatitis C virus, or with which the HCV is in any way associated.

According to a third aspect the present invention provides the use of a compound or derivative according to the first aspect in the manufacture of a medicament for the treatment or prevention of hepatitis C or a related condition.

A fourth aspect of the invention provides a pharmaceutical composition which includes one or more compounds or derivatives according to the first aspect.

The composition may also include pharmaceutically acceptable adjuvants such as carriers, buffers, stabilisers and other excipients. It may additionally include other therapeutically active agents, in particular those of use in treating or preventing hepatitis C or related conditions.

35 The pharmaceutical composition may be in any suitable

5 form, depending on the intended method of administration.
It may for example be in the form of a tablet, capsule or
liquid for oral administration, or of a solution or
suspension for administration parenterally.

10 According to a fifth aspect of the invention, there is
provided a method of inhibiting HCV NS3 protease
activity, and/or of treating or preventing hepatitis C or
a related condition, the method involving administering
15 to a human or animal (preferably mammalian) subject
suffering from the condition a therapeutically or
prophylactically effective amount of a composition
according to the fourth aspect of the invention, or of a
compound or derivative according to the first aspect.
"Effective amount" means an amount sufficient to cause a
20 benefit to the subject or at least to cause a change in
the subject's condition.

The dosage rate at which the compound, derivative or
composition is administered will depend on the nature of
25 the subject, the nature and severity of the condition,
the administration method used, etc. Appropriate values
can be selected by the trained medical practitioner.
Preferred daily doses of the compounds are likely to be
of the order of about 1 to 100 mg. The compound,
30 derivative or composition may be administered alone or in
combination with other treatments, either simultaneously
or sequentially. It may be administered by any suitable
route, including orally, intravenously, cutaneously,
subcutaneously, etc. Intravenous administration is
35 preferred. It may be administered directly to a suitable

5

site or in a manner in which it targets a particular site, such as a certain type of cell - suitable targeting methods are already known.

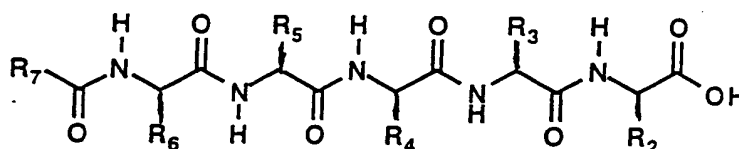
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A sixth aspect of the invention provides a method of preparation of a pharmaceutical composition, involving admixing one or more compounds or derivatives according to the first aspect of the invention with one or more pharmaceutically acceptable adjuvants, and/or with one or more other therapeutically or prophylactically active agents.

15

The compounds themselves may be prepared by reacting a compound of formula 4:

20



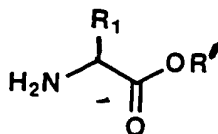
Formula 4

25

optionally in a protected form,

with an appropriate amine co-reactant (depending on the intended nature of R_1 and X in the final compound), examples of which include;

30

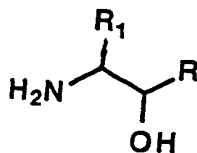


FORMULA 5

35

5 (for X = OH, as in compounds 1a, 1b and 1f), R' being a protecting group;

10

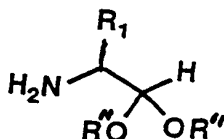


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FORMULA 6

(for X = R or H, eg, as in compounds 1c, 1d, 1e, 1g, 1h 1j, 1k, 1L, or 1m); and

20



25

FORMULA 7

(for X = H, as in compound 1c, R'' being a lower alkyl group such as methyl or ethyl).

30

Compounds of formula 1 having m=1 may be produced using homologs of the above compounds 5, 6 and 7 including an additional CH₂ group at the appropriate position.

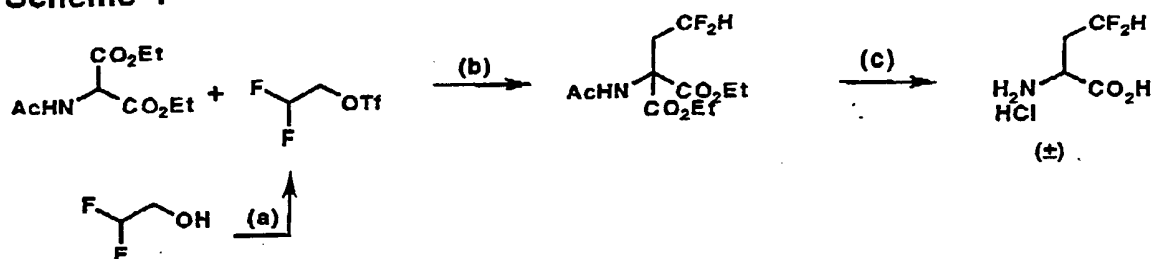
35 Compounds of formulae 5, 6 and 7 may be used as racemates

5 or, alternatively, as individual D- or L-isomers. When a
racemate is used subsequent separation of product
diastereomers may be desirable.

10 In each case, the reaction can be carried out using
standard methods of peptide synthesis. In the case of
formula 6, oxidation of the alcohol to a ketone is also
needed. In all cases, protecting groups may need to be
removed, for instance under mildly acidic or basic
conditions, to reach the final product.

15 A preferred compound of formula 5 is racemic 4,4-
difluoro-2-aminobutyric acid. One possible scheme for
the preparation of this compound is set out below in
scheme 1

20 **Scheme 1^a**

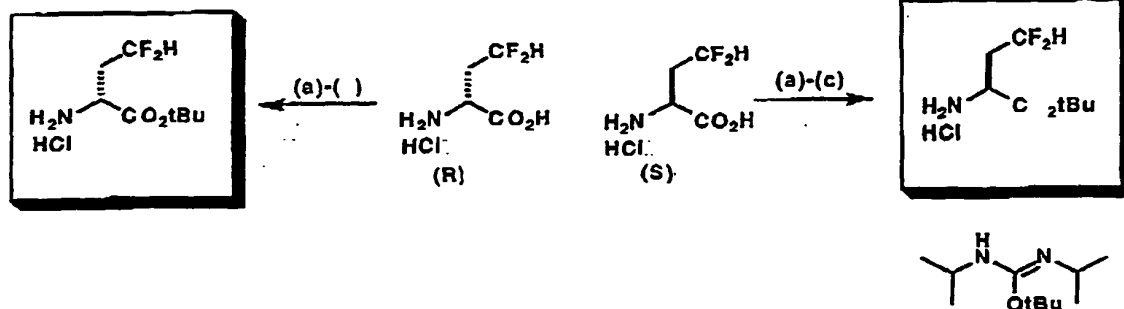


25 ^aReagents: (a) TiF₄·O, CH₂Cl₂, Et₃N; (b) KOtBu, THF, Δ; (c) 6N HCl, Δ

The individual R- and S- enantiomers of 4,4-difluoro-2-
aminobutyric acid may be prepared from D- and L- aspartic
acid, respectively using the method described by Winkler
et al in Synthesis (1996), 1419-1421. The carboxylic
30 acid group of these compounds may be protected, for
instance by formation of t-butyl esters as shown below in
scheme 2

35

Scheme 2^a

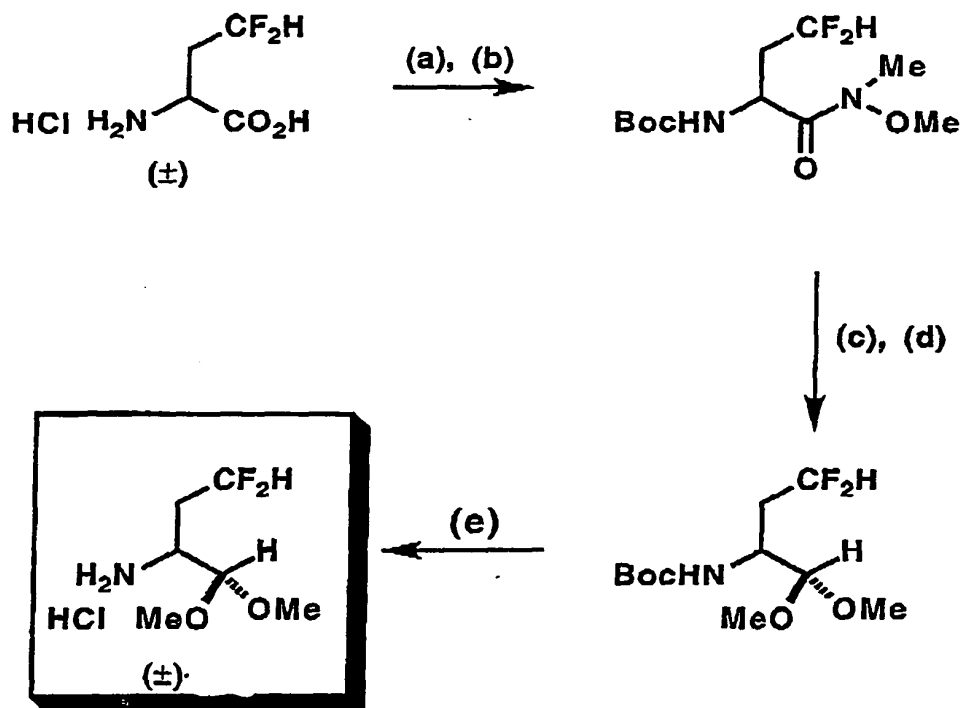


^aReagents: (a) CbzOSuc, Na₂CO₃, dioxan; (b) 10, CH₂Cl₂; (c) H₂, Pd/C, ether/HCl

10

One example of a racemic diacetal of formula 7 may be prepared as outlined below in scheme 3 which begins with racemic 4,4-difluoro-2-aminobutyric acid.

Scheme 3^a



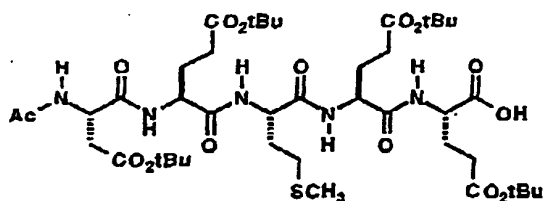
^aReagents: (a) Boc₂O; (b) NH(OMe)Me.HCl, EDCI.HCl, HOBT, iPr₂NEt; (c) DIBAL, THF, -78°; (d) HC(OMe)₃, TsOH; (e) HCl, MeOH, 0°-rt

5 Compounds of formula 4 may be generated wholly or partly by chemical synthesis, and in particular can be prepared according to known peptide synthesis methods.

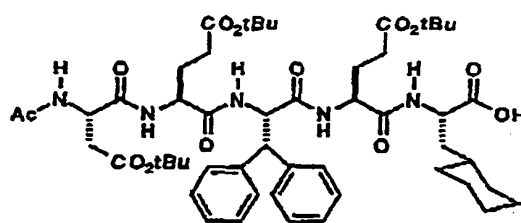
10 Preferably, all five amino acids, or amino acid analogs in the compound of formula 4 are L-isomers.

Preferably, the compound of formula 4 for reaction with a compound of formula 5, 6 or 7 will be in protected form. For instance, any carboxylic acid groups other than that at the C terminus may preferably be protected, for instance as esters, eg as tertiary butyl esters. Examples of two highly preferred protected pentapeptides are set out below and labelled (A) and (B)

20



(A)



(B)

25

The invention provides, according to a seventh aspect, a method as described above for preparing a compound according to the first aspect.

30

An eighth aspect of the invention provides a compound of formula 4 (in which the definitions of R_2 - R_7 are as for compounds of formula 1), as an intermediate in a preparation method according to the seventh aspect.

35

5 Examples

Embodiments of the invention are described below by way of example only

10 (1) Synthesis

HPLC Conditions: Reversed phase analytical HPLC was performed on a Waters Symmetry C18 column (150 x 3.9 mm, 5 μ m), flow rate 1 mL/min, using H₂O/0.1% TFA (A) and CH₃CN/0.1% TFA (B) as eluents; detection at 220 nm with a Waters 996 PDA detector. Gradient 1: linear, 90 A- 20% A 8 min, then in 2 min to 0% A, then isocratic. Gradient 2: linear, 70 - 40% A 10 min. . Gradient 3: linear, 90 - 70% A 10 min. Preparative HPLC was conducted on a Waters Symmetry C18 column (150 x 19 mm, 7 μ m) or a Waters Prep Nova-Pak HR C18 cartridge (40 x 100 mm, 6 μ m) using H₂O/0.1% TFA (A) and CH₃CN/0.1% TFA (B) as eluents; detection at 220 nm with a Waters 486 absorbance detector.

25

EXAMPLE 1

Synthesis of compound 1a

30 i) 1,1-Difluoro-2-trifluoromethanesulfonyloxyethane

Triflic anhydride (120 g, 0.427 mol) was dissolved in anhydrous dichloromethane (70 mL) and cooled to -60° C. A solution of triethylamine (59.5 mL, 0.427 mol) and difluoroethanol (35 g, 0.427 mol) in dichloromethane (70

5

mL) was added slowly, so that the internal temperature did not exceed -50°C . After complete addition the resulting yellow solution was allowed to reach room temperature. Dichloromethane was distilled off under atmospheric pressure, and the remaining liquid fractioned under reduced pressure (70 - 80 mbar), using a 20 cm Vigreux column to give the title sulfonate (86.2 g, 94%) (b.p.: $58 - 60^{\circ}\text{C}$). $^1\text{H-NMR}$ (CDCl_3) 4.58 (dt, $J = 3.6, 12.8$ Hz, 2 H, CH_2), 6.05 (tt, $J = 3.6, 54$ Hz, 1 H, CHF_2); $^{19}\text{F-NMR}$ (CDCl_3) δ -74.6 (s), -127 (s).

15

ii) Diethylacetamido-2-(2',2'-difluoroethyl) malonate

Diethyl acetamido malonate (35.8 g, 0.165 mol) was dissolved in anhydrous THF (300 mL) and treated with potassium tert-butanolate (18.5 g, 0.165 mol) under vigorous stirring. The resulting suspension was refluxed for 1.5 h, and the above sulfonate (40 g, 0.187 mol) was added carefully via syringe to the refluxing suspension. The solution became homogeneous and was refluxed for another 3h. The solution was concentrated, and the residue dissolved in ethyl acetate and washed with hydrochloric acid (0.5 N, 2x), water (2x), saturated aqueous NaHCO_3 , sodium hydroxide (1 N, 1x) and brine. Drying (Na_2SO_4) and evaporation left an orange oil, which was dissolved in diethyl ether (250 mL). The flask was kept at -20°C overnight. 32.6 g (70%) of a colourless solid was collected; mp $72 - 73^{\circ}\text{C}$. $^1\text{H-NMR}$ (CDCl_3) δ 1.26 (t, $J = 7.1$ Hz, 6 H, CH_3), 2.05 (s, 3 H, COCH_3), 2.98 (dt, $J = 4.7, 16.5$ Hz, 2 H, CH_2CHF_2), 4.27 (q, $J = 7.1$ Hz, 4 H, CH_2), 5.85 (tt, $J = 4.7, 55.8$ Hz, 1 H, CHF_2), 6.90

5 (bs, 1 H, NH); ^{13}C -NMR (CDCl_3) δ 13.8, 22.9, 36.8 (t, J = 22.6 Hz), 62.8, 63.1, 115.2 (t, J = 239 Hz), 167.0, 169.7; ^{19}F -NMR (CDCl_3) δ -116.8 (s); MS m/z 282 ($\text{M}^+ + \text{H}$).

10 iii) (+)-2-Amino-4,4-difluorobutanoic acid hydrochloride

The malonate prepared above (32 g, 0.114 mol) was refluxed in 500 mL hydrochloric acid (6 N) overnight. The aqueous phase was extracted with diethyl ether and then evaporated to give the title compound (19.9 g;
15 quantitative yield) as a colourless solid; mp 164 - 165 °C. ^1H -NMR (D_2O) δ 2.35 - 2.70 (m, 2 H, CH_2), 4.27 (dd, J = Hz, 1 H, CH), 6.19 (tt, J = Hz, 1 H, CHF_2); ^{13}C -NMR (D_2O) δ 34.0 (t, J = 22.2 Hz), 48.2, 115.7 (t, J = 238 Hz), 171.4; ^{19}F -NMR (D_2O) δ -112.7 (d, 287 Hz), -114.2 (d, 287
20 Hz); MS m/z 149 ($\text{M}^+ + \text{H}$).

iv) (R)-tert-Butyl-2-amino-4,4-difluoro butanoate hydrochloride

25 1.5 g (10.78 mmol) of (R) 2-Amino-4,4-difluoro butanoic acid (prepared as described in Winkler et al, Synthesis 1419 1996) was dissolved in aqueous half saturated Na_2CO_3 (50 mL) and cooled to 0 °C. A solution of
(benzyloxycarbonyloxy)succinimide (2.69 g, 10.78 mmol) in
30 dioxane (50 mL) was added dropwise over 30 min. The resulting suspension was stirred overnight at room temperature. After evaporation of the dioxane under reduced pressure water (20 mL) and EtOAc (150 mL) were added. The aqueous phase was brought to pH 2 by addition

5 of 1 N HCl, the organic phase was separated, washed with brine and dried. Evaporation gave 2.85 g (97%) of a colourless oil.

10 This material (950 mg; 3.55 mmol) was dissolved in dichloromethane (15 mL) and *N,N'*-isopropyl-*O*-*t*butyl isourea (1.42 g, 7.10 mmol) was added dropwise. The solution was brought to gentle reflux. After 8 h another 1.42 g of the isourea was added and reflux was continued overnight. The diisopropylurea was removed by filtration,
15 and the residue purified by flash chromatography (petroleum ether/ethyl acetate 10 : 1) to give a colourless oil (844 mg; 72%). ¹H-NMR (DMSO-*d*₆) δ ¹H-NMR (DMSO-*d*₆) δ 1.38 (s, 9 H), 2.14 - 2.28 (m, 2 H), 4.08 (m, 1 H), 5.03 (d, *J* = 12.6 Hz, 1 H),), 5.06 (d, *J* = 12.6
20 Hz, 1 H), 6.10 (tt, *J* = 4.7, 56.2 Hz, 1 H), 7.27 - 7.39 (m, 5 H), 7.79 (d, *J* = 8.1 Hz, 1 H); ¹³C-NMR (DMSO-*d*₆) δ 27.4, 34.9 (t, *J* = 22.5 Hz), 49.5, 65.5, 81.2, 115.9 (t, *J* = 238 Hz), 127.7, 127.8, 128.3, 136.7, 155.8, 169.8; ¹⁹F-NMR (DMSO-*d*₆) δ -115.1 (d, *J* = 283 Hz), -115.8 (d, *J* =
25 283 Hz); MS *m/z* 330 (*M*⁺ + H).

300 mg (0.91 mmol) of this material were hydrogenated over 10% palladium-on-charcoal in methanol (10 mL). After 5h, the catalyst was removed by filtration, then some
30 ethyl acetate and a 1 N solution of hydrochloric acid in diethyl ether (1.37 mL) were added. After evaporation in vacuo the title compound (203 mg; 96%) was obtained as an off-white solid; mp 153 - 154 °C; ¹H-NMR (DMSO) δ 1.44 (s, 9 H), 2.38 - 2.50 (m, 2 H), 4.03 (t, *J* = 6.2 Hz, 1 H),

5 6.35 (tt, $J = 4.3$, 55.6 Hz, 1 H), 8.85 (bs, 3H); ^{13}C -NMR
(DMSO- d_6) δ 27.3, 34.3 (t, $J = 23.3$ Hz), 47.6, 83.4, 114.9
(t, $J = 238$ Hz), 167.0; ^{19}F -NMR (DMSO- d_6) δ -114.5 (d, $J =$
285 Hz), -115.3 (d, $J = 285$ Hz); MS m/z 196 ($M^+ + H$).

10

v) (S)-tert.-Butyl-2-amino-4,4-difluoro-butanoate
 hydrochloride

Using the procedure described above for the (R)-
15 enantiomer, the title compound was obtained as an off-
white powder; mp 152 - 153 °C (MeOH, Et₂O, pentane); α_D
+5.1° ($c = 1.0$, anhydrous MeOH). ^1H -NMR (DMSO- d_6) δ 1.44
(s, 9 H), 2.36 - 2.50 (m, 2 H), 4.05 (bs, 1 H), 6.31 (tt,
 $J = 4.5$, 55.6 Hz, 1 H), 8.71 (bs, 3H); ^{13}C -NMR (DMSO- d_6) δ
20 27.3, 34.3 (t, $J = 23.3$ Hz), 47.6, 83.5, 114.9 (t, $J =$
238 Hz), 167.1; ^{19}F -NMR (DMSO- d_6) δ -114.4 (d, $J = 285$
Hz), -115.2 (d, $J = 285$ Hz); MS m/z 196 ($M^+ + H$).

25 vi) (+)-(2-N-(tert-Butoxycarbonyl)-amino)-4,4-difluoro-
 butyric N-methyl-O-methylcarboxamide

1.0 g (5.7 mmol) of (+)-2-amino-4,4-difluoro butanoic
acid hydrochloride was converted to its Boc derivative
30 using di-tert.-butyl dicarbonate (1.24 g, 5.7 mmol).
After extractive workup 1.16 g (85%) of a colourless
solid was obtained, which was used without further
purification; mp 127 - 129 °C. ^1H -NMR (DMSO- d_6) δ 1.37 (s,
9 H, t-Bu), 2.15 (m, 2 H, CH₂), 4.03 (m, 1 H, α -CH), 6.07

5

(tt, $J = 4.5$, 56 Hz, 1 H, CHF_2), 7.30 (d, $J = 8.5$ Hz, 1 H, NH), 12.80 (bs, 1 H, COOH); ^{13}C -NMR ($\text{DMSO}-d_6$) δ 28.0, 35.0 (t, $J = 22$ Hz), 48.4, 78.3, 116.0 (t, $J = 238$ Hz), 155.3, 172.5; ^{19}F -NMR ($\text{DMSO}-d_6$) δ -115.0 (d, $J = 282$ Hz), -115.7 (d, $J = 282$ Hz); MS m/z 240 ($\text{M}^+ + \text{H}$).

10

To a solution of the Boc derivative prepared above (1.59 g, 6.65 mmol), EDC•HCl (1.40 g, 7.32 mmol) and HOBT (1.08 g, 7.98 mmol) in anhydrous dichloromethane (30 mL) was added a solution of *N,O*-dimethylhydroxylamine

15

hydrochloride (714 mg, 7.32 mmol) and diisopropylethylamine (1.74 mL, 9.98 mmol) in dichloromethane (20 mL) at 0 °C. After stirring at room temperature for 3 days, some dichloromethane was removed under reduced pressure. The resulting solution was

20

diluted with ethyl acetate (150 mL) and washed successively with 1 N HCl (2x), sat. aqueous NaHCO_3 (2x) and brine. The organic extract was dried (Na_2SO_4) and concentrated in vacuo to give the title compound (1.81 g; 96%) of as a colourless solid. A small sample was

25

recrystallized to give: mp 81 -82 °C. ^1H -NMR (CDCl_3) δ 1.44 (s, 9 H, t-Bu), 1.93 - 2.44 (m, 2 H, CH_2), 3.23 (s, 3 H, NCH_3), 3.76 (s, 3 H, OCH_3), 4.84 (m, 1 H, $\alpha\text{-CH}$), 5.39 (bd, $J = 9.0$ Hz, 1 H, NH), 5.95 (ddt, $J = 3.6, 5.8, 56.0$ Hz, 1 H, CHF_2); ^{13}C -NMR (CDCl_3) δ 28.3, 32.3, 37.6 (t, $J = 22$ Hz), 46.3, 61.7, 80.2, 115.3 (t, $J = 239$ Hz), 155.3, 171.2; ^{19}F -NMR (CDCl_3) δ -114.6 (d, $J = 287$ Hz), -115.5 (d, $J = 287$ Hz); MS m/z 283 ($\text{M}^+ + \text{H}$).

30

(vii) (+)-2-(*N*-tert.-Butoxycarbonyl)amino-4,4-

To a solution of the above urethane (4.89 g, 17.32 mmol) in tetrahydrofuran (100 mL) was added neat diisobutylaluminum hydride (6.79 mL, 38.11 mmol) dropwise at -78 °C. The solution was stirred for 2.5 h at this temperature, then methanol (5 mL) was added dropwise and the cooling bath removed. The solution was diluted with ethyl acetate (500 mL) and then washed successively with ice-cold 1 N HCl (150 mL, 3x), 2 N aqueous Rochelle's salt (150 mL) and brine (2x). Drying of the organic extract (Na₂SO₄) and evaporation in vacuo gave 3.47 g (90%) of (±)-2-(N-tert.-Butoxy carbonyl)-amino-4,4-difluoro butyraldehyde as an opaque oil, which was used in the next step without further purification. ¹H-NMR (CDCl₃) δ 1.47 (s, 9 H, t-Bu), 2.25 (m, 1 H, CH₂), 2.55 (m, 1 H, CH₂), 4.31 (m, 1 H, α-CH), 5.33 (bs, 1 H, NH), 6.03 (dt, J = 6.0, 56 Hz, 1 H, CHF₂), 9.60 (s, 1 H, CHO);

1.8 g (8.06 mmol) of the crude aldehyde were converted into the dimethylacetal using trimethylorthoformate (12.4 mL, 112.9 mmol) and p-toluenesulfonic acid (154 mg, 0.81 mmol) in anhydrous methanol (30 mL). After stirring overnight at room temperature, TLC (petroleum/ethyl acetate 2:1) indicated complete consumption of the aldehyde. Saturated aqueous NaHCO₃ was added and the methanol evaporated under reduced pressure. The residue was dissolved with ethyl acetate (200 mL) and washed successively with saturated aqueous NaHCO₃ and brine. Drying (Na₂SO₄) and evaporation left an oil which was

5 purified by flash chromatography (160 g silica gel,
petroleum/ethyl acetate 4 : 1, containing 0.5%
triethylamine), to give the title compound (1.44 g; 66%)
as a colourless solid; mp 61 -62 °C. ¹H-NMR (CDCl₃) δ 1.48
(s, 9 H, t-Bu), 1.86 - 2.05 (m, 1 H, CH₂), 2.09 - 2.27 (m,
10 1 H, CH₂), 3.44 (s, 3 H, OCH₃), 3.45 (s, 3 H, OCH₃), 3.99
(m, 1 H, α-CH), 4.25 (d, J = 3.0 Hz, 1 H, CH(OMe)₂), 4.76
(bd, J = 8.0 Hz, 1 H, NH), 5.96 (ddt, J = 4.0, 5.4, 56.6
Hz, 1 H, CHF₂); ¹³C-NMR (CDCl₃) δ 28.3, 34.4 (t, J = 22 Hz,
CH₂CHF₂), 47.6, 55.9, 56.5, 79.8, 105.6, 116.3 (t, J = 238
15 Hz, CH₂CHF₂), 155.5; ¹⁹F-NMR (CDCl₃) δ -114.6 (d, J = 284
Hz), -115.5 (d, J = 284 Hz); MS m/z 270 (M⁺ + H).

viii) (+)-2-Amino-4,4-difluorobutyraldehyde
dimethylacetal hydrochloride

20

To 300 mg (1.11 mmol) of the above acetal was added a
solution of gaseous HCl in anhydrous methanol (10% HCl by
weight) at 0 °C. The solution was stirred at this
temperature for 30 min, then 20 min at ambient
25 temperature. Evaporation and drying under high vacuum
gave the title compound (200 mg) as a yellow oil.

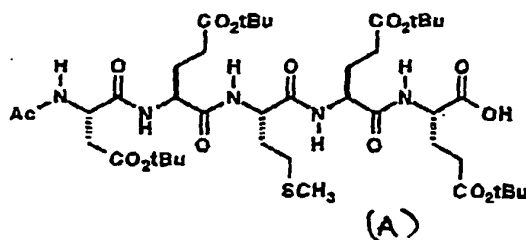
(ix) Ac-Asp-Glu-Met-Glu-Glu-Difluoroaminobutyric acid
(1a)

30

The protected pentapeptide shown below (ac-tert-butyl-
asp-tert-butyl-glu-met-tert-butyl-glu-tert-butyl-glu) was
employed in this example

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10 30 mg pentapetide (0.03 mmol) was dissolved in
dichloromethane (0.5 mL) and cooled to 0 °C. N-Ethyl, N'-
(dimethylamino)propylcarbodiimide hydrochloride
(EDCI.HCl) (6.3 mg, 0.033 mmol) and hydroxybenzotriazole
15 (s)-tert-butyl-2-amino-4,4-difluoro-butanoate
hydrochloride (from v, above) (10.4 mg, 0.045 mmol) and
diisopropylethylamine (11 µL, 0.06 mmol). The resulting
solution was stirred overnight at room temperature, then
taken into ethyl acetate (50 mL) and washed successively
20 with 1 N HCl (2x 25 mL), saturated aqueous NaHCO₃ (2 x 20
mL), and brine. Drying (Na₂SO₄) and evaporation gave a
solid which was immediately treated with a solution of
trifluoroacetic acid, dichloromethane and water
(60/30/10, v/v/v; 10 mL). After 30 min at room
25 temperature the solvents were evaporated in vacuo and the
remaining solid separated by preparative HPLC (Waters
Symmetry column). Flow 17 mL/min; Gradient : linear, 90%
A, 3 min isocratic, in 15 min to 75% A; 7 mg of crude per
injection. The product, compound 1a (RT 10.4 min); 12 mg
30 (50%) was obtained as a colourless solid after
lyophilization.

¹H-NMR (DMSO-d₆) δ 1.73 - 1.95 (m, 8 H), 1.83 (s, 3 H),
2.02 (s, 3 H), 2.19 - 2.30 (m, 8 H), 2.35 - 2.48 (m, 3 H),
35 2.61 (dd, J = 5.2, 11.7 Hz, 1 H), 4.14 - 4.26 (m, 3 H),

4.29 (m, 1 H), 4.36 (m, 1 H), 4.50 (dd, $J = 5.4, 7.7$ Hz, 1 H), 6.05 (ddt, $J = 4.6, 51.6$ Hz, 1 H), 7.92 (d, 1 H, $J = 8.4$ Hz, 1 H), 7.96 (d, 1 H, $J = 8.2$ Hz, 1 H), 7.99 (m, 2 H),), 8.18 (d, 1 H, $J = 7.5$ Hz, 1 H), 8.33 (bd, 1 H, $J = 7.0$ Hz, 1 H), 11.9 - 12.4 (bs, 5 H); ^{19}F -NMR (DMSO- d_6) δ -115.0 (d, $J = 282$ Hz), -115.8 (d, $J = 284$ Hz); MS m/z 815 ($\text{M}^+ + \text{H}$).

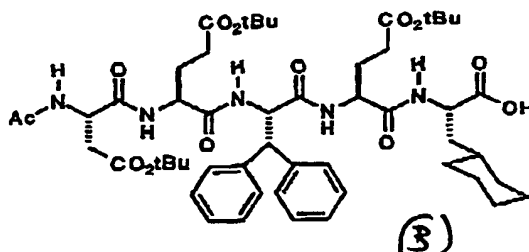
EXAMPLE 2

Synthesis of compound 1b¹

Ac-Asp-Glu-Diphenylalanine-Glu- β -Cyclohexylala-Difluoroaminobutyric acid (1b¹)

In this example, (S)-tert-butyl-2-amino-4,4-difluorobutanoate was used in the preparation of a first diastereomer of compound 1b.

This example, and also examples 3 and 4 below, employed the protected pentapeptide shown below (ac-tert-butyl-aspart-tert-butyl-glu-diphenylala-tert-butyl glu- β -cyclohexyl-ala)



5 50 mg pentapeptide (0.05 mmol) was dissolved in DMF (0.5 mL) and cooled to 0 °C. HATU and solid (S)-tert-butyl-2-amino-4,4-difluoro-butanoate hydrochloride (from v, above) were added, followed by 2,6-lutidine (24 µL, 0.2 mmol). The reaction was allowed to reach room temperature and stirred for 3 h. Analytical HPLC (gradient 1) indicated incomplete conversion of the pentapeptide (~30% remaining, RT 10.4 min, gradient 1, product 11.9 min). After another 2 h the mixture was taken into ethyl acetate (100 mL) and washed successively with 1 N HCl, 10 (2x 50 mL), saturated aqueous NaHCO₃ (2 x 50 mL), and brine. Drying (Na₂SO₄) and evaporation gave a light yellow solid which was immediately deprotected with a solution of trifluoroacetic acid, dichloromethane and water 15 (60/30/10, v/v/v; 10 mL). After 30 min at room temperature the solvents were evaporated in vacuo and the 20 remaining solid separated by preparative HPLC (Waters Symmetry column). Flow 17 mL/min; Gradient : linear, 68% A, 3 min isocratic, in 17 min to 65% A; 6 mg of crude per injection. The first peak was deprotected pentapeptide (RT 25 11.6 min), the second the desired product compound 1b (RT 12.2 min); 11 mg (23%) of a colourless solid after lyophilization.

¹H-NMR (DMSO-d₆) δ 0.76-0.95 (m, 2 H), 1.08 - 1.32 (m, 4 30 H), 1.32 - 1.41 (m, 1 H), 1.42 - 1.51 (m, 1 H), 1.53-1.80 (m, 9 H), 1.83 (s, 3 H), 1.97 - 2.35 (m, 6 H), 2.38 - 2.50 (m, 2 H), 4.04 - 4.13 (m, 2 H), 4.13 - 4.21 (m, 1 H), 4.27 - 4.37 (m, 1 H), 4.38 (d, J = 10.3 Hz, 1 H), 4.47 (m, 1 H), 5.19 (app. t, J = 9.5 Hz, 1 H), 6.04 (ddt, 35 J = 4.0, 5.7, 56.2 Hz, 1 H), 7.05-7.33 (m, 10 H), 7.75

5 (d, 1 H, $J = 7.3$ Hz, 1 H), 7.79 (d, 1 H, $J = 8.0$ Hz, 1 H), 7.89 (d, 1 H, $J = 8.1$ Hz, 1 H),), 7.96 (d, 1 H, $J = 7.6$ Hz, 1 H), 8.10 (d, 1 H, $J = 7.0$ Hz, 1 H), 8.10 - 8.12 (bs, 1 H); MS m/z 929 ($M^+ - H$).

10

EXAMPLE 3

Synthesis of compound 1b²

15 In this example, (R)-tert-butyl-2-amino-4,4-difluorobutanoate hydrochloride (from iv, above) was used to prepare the alternative diastereomer to that prepared in example 2. The method was as described in example 2.

20 After 3 h analytical HPLC indicated only minor amounts of the protected pentapeptide. After workup the crude product was deprotected as described in example 2 and separated by preparative HPLC (Waters Symmetry column). Flow 17 mL/min; Gradient : linear, 70% A, 3 min
25 isocratic, in 12 min to 40% A; 6 mg of crude per injection. 22 mg (47%) of 17 (RT 9.2 min) as a colourless solid were obtained after lyophilization.

30 ¹H-NMR (DMSO-d₆) δ 0.77-0.91 (m, 2 H), 1.06 - 1.25 (m, 4 H), 1.29 - 1.36 (m, 1 H), 1.37 - 1.44 (m, 1 H), 1.52-1.80 (m, 9 H), 1.82 (s, 3 H), 1.99 - 2.13 (m, 4 H), 2.16 - 2.33 (m, 2 H), 2.42 (dd, $J = 8.8, 16.6$ Hz, 1 H, β -CH₂ Asp), 2.49 (under DMSO, m, 1 H), 4.08 (m, 2 H), 4.21 (m, 1 H), 4.33 (m, 1 H), 4.37 (d, $J = 10.3$ Hz, 1 H), 4.47 (m, 1 H), 5.21 (app. t, $J = 9.4$ Hz, 1 H), 5.99 (dt, $J = 4.6,$

35

5 56.3 Hz, 1 H), 7.05-7.40 (m, 10 H), 7.65 (d, 1 H, $J = 7.7$
Hz, 1 H), 7.78 (d, 1 H, $J = 7.9$ Hz, 1 H), 7.87 (d, 1 H, J
= 8.4 Hz, 1 H),), 7.96 (d, 1 H, $J = 7.8$ Hz, 1 H), 8.14
(d, 1 H, $J = 7.7$ Hz, 1 H), 8.30 (d, 1 H, $J = 8.10$ Hz, 1
H), 11.90 - 12.30 (bs, 4 H); MS m/z 929 ($M^+ - H$).

10

EXAMPLE 4

Synthesis of compound 1c

15 220 mg of the protected pentapeptide (Ac-tert-butyl-asp-
tert-butyl-glu-diphenylalanine-tert-butyl-glu- β -
cyclohexylala) (0.225 mmol) were dissolved in 1 mL
chloroform. EDC•HCl (52 mg, 0.27 mmol) and HOBT (61 mg,
0.45 mmol) were added and the solution cooled to 0 °C.
20 The acetal, (\pm 2-amino-4,4-difluorobutyraldehyde
dimethylacetal hydrochloride (from viii above) (80 mg,
0.39 mmol) was dissolved in chloroform (0.8 mL)
containing Hünig's base (0.47 mmol, 0.082 mL) and the
resulting solution was added via syringe to the
25 pentapeptide. Another 0.3 mL chloroform was used to rinse
flask and syringe. The cooling bath was removed after 10
min and the orange solution stirred for 3 h. Analytical
HPLC indicated complete conversion of the pentapeptide.
The reaction was taken into a mixture of ethyl acetate
30 and dichloromethane (150 mL, 3 : 1) and washed
successively with 0.1 M aqueous KHSO_4 , (3x 80 mL), water
(2x 100 mL), saturated aqueous NaHCO_3 , and brine (2x 100
mL). Drying (Na_2SO_4) and evaporation gave a brown solid
which was immediately deprotected with a solution of
35 trifluoroacetic acid, dichloromethane and water (60/35/5,

5 v/v/v; 50 mL). After 30 min at room temperature the
solvents were evaporated in vacuo and the remaining brown
solid (252 mg) comprising a mixture of diastereomers was
separated by preparative HPLC (Nova-Pak Prep column).
Flow 40 mL/min; Gradient : linear, 70% A, 2 min
10 isocratic, in 18 min to 60% A; 20 mg of crude per
injection.

First fraction: RT: 9.4 min, 54 mg (26%) of a colourless
powder after lyophilization; 1 diastereomer, 94% pure by
15 analytical HPLC (gradient 1, 6.77 min; gradient 2, 6.45
min). In the ¹H-NMR 10 - 20% of the aldehyde was visible
as its hydrate. Addition of water gave a ratio of
aldehyde to hydrate of 1 : 9. Only data for the aldehyde
are reported. ¹H-NMR (DMSO-d₆) δ 0.77 - 0.94 (m, 2 H),
20 1.05 - 1.31 (m, 4 H), 1.32 - 1.50 (m, 2 H), 1.52 - 1.78
(m, 9 H), 1.82 (s, 3 H), 1.95 - 2.15 (m, 6 H), 2.36 -
2.46 (m, 2 H, β-CH₂ Asp), 4.00 - 4.06 (m, 2 H, α-CH
difluoro, Glu),), 4.12 - 4.23 (m, 2 H, α-CH Cha, Glu),
4.39 (d, *J* = 10.3 Hz, 1 H, β-CH Dif), 4.47 (m, 1 H, α-CH
25 Asp), 5.19 (app. t, *J* = 9.4 Hz, 1 H, α-CH Dif), 6.10 (dt,
J = 4.6, 56.0 Hz, 1 H, CHF₂), 7.05 - 7.38 (m, 10 H), 7.75
(d, *J* = 7.3 Hz, 1 H, NH difluoro), 7.81 (d, *J* = 6.9 Hz, 1
H, NH Cha), 7.86 (d, *J* = 8.0 Hz, 1 H, NH Dif), 8.10 (m, 2
H, NH Glu), 8.40 (d, *J* = 7.2 Hz, 1 H, NH Asp), 9.26 (s, 1
30 H, CHO), 11.50 - 12.50 (bs, 3 H, COOH); MS *m/z* 915 (*M*⁺ +
H). HRMS (Fab) (C₄₄H₅₆F₂N₆O₁₃) calc. 914.3873, fd.

Second fraction: RT: 12.2 min, 42 mg (20%), colourless
powder after lyophilization;

35

5 ¹H-NMR (DMSO-d₆) δ 0.76 - 0.94 (m, 2 H), 1.05 - 1.30 (m, 4
H), 1.32 - 1.50 (m, 2 H), 1.52 - 1.78 (m, 9 H), 1.83 (s,
3 H), 1.95 - 2.15 (m, 6 H), 2.25 - 2.45 (m, 2 H), 3.98 -
4.12 (m, 2 H),), 4.15 - 4.23 (m, 2 H), 4.35 - 4.51 (m, 2
H), 5.15 - 5.19 (m, 1 H), 6.06 (dt, *J* = 4.5, 56.1 Hz, 1
10 H), 7.07 - 7.38 (m, 10 H), 7.58 (d, *J* = 7.5 Hz, 1 H),
7.60 - 8.12 (m, 4 H), 8.43 (bs, 1 H), 9.32 (s, 1 H, CHO),
11.90 (bs, 3 H, COOH); MS *m/z* 915 (M⁺ + H).

Using analogous methods, compounds 1d to 1m were also
15 produced.

2. INHIBITION OF NS3 PROTEASE

20 The ability of the compounds to inhibit NS3 protease was
evaluated using an NS3/4A complex comprising the NS3
protease domain and a modified form of the NS4A peptide,
Pep 4AK [KKKGSVVIVGRIILSGR(NH₂)]. As substrate, a
substrate peptide 4AB [DEMEECASHLPYK] based on the
25 sequence of the NS4A/NS4B cleavage site of the HCV
polyprotein, was used

Cleavage, assays were performed in 57 μl 50 mM Hepes
pH7.5, 1 % CHAPS, 15 % glycerol, 10 mM DTT (buffer A), to
30 which 3 μl substrate peptide were added. As protease co-
factor a peptide spanning the central hydrophobic core
(residues 21-34) of the NS4A protein, Pep4AK
[KKKGSVVIVGRIILSGR(NH₂)] was used. Buffer solutions
containing 80 μM Pep4AK were preincubated for 10 minutes
35 with 10-200 nM protease and reactions were started by

5 addition of substrate. Six duplicate data points at
different substrate concentrations were used to calculate
kinetic parameters. Incubation times were chosen in
order to obtain <7% substrate conversion and reactions
were stopped by addition of 40 μ l 1 % TFA. Cleavage of
10 peptide substrates was determined by HPLC using a Merck-
Hitachi chromatograph equipped with an autosampler. 80
 μ l samples were injected on a Lichrospher C18 reversed
phase cartridge column (4 x 74mm, 5 μ m, Merck) and
fragments were separated using a 10-40 % acetonitrile
15 gradient a 5%/min using a flow rate of 2.5ml/min. Peak
detection was accomplished by monitoring both the
absorbance at 220nm and tyrosine fluorescence (λ_{ex} = 260
nm, λ_{em} = 305nm). Cleavage products were quantitated by
integration of chromatograms with respect to appropriate
20 standards. Kinetic parameters were calculated from
nonlinear least-squares fit of initial rates as a
function of substrate concentration with the help of a
Kaleidagraph software, assuming Michaelis-Menten
kinetics.

25 K_i values of peptide inhibitors were calculated from
substrate titration experiments performed in the presence
of increasing amounts of inhibitor. Experimental data
sets were simultaneously fitted to eq.1 using a
multicurve fit macro with the help of a Sigmaplot
30 software:

$$V = (V_{max}S) / (K_m(1+K_i/I)+S); \quad (eq.1)$$

Alternatively, K_i values were derived from IC50 values,
35 calculated using a four-parameter logistic function,

5 according to eq.2:

$$\text{IC}_{50} = (1 + S/K_m) K_i \quad (\text{eq.2})$$

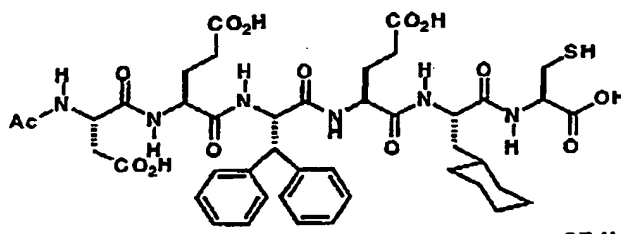
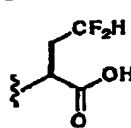
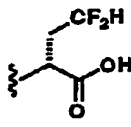
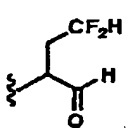
10 Results for the compounds synthesized in Examples 1 to 4 above are tabulated below in Tables 1 and 2. In Table 2, the data shown for compound 1c are for the more active diastereomer.

Table 1. Inhibition of NS3 Protease^a

	cmpd #	K _i
		0.7
	1a	1.5

^aK_i values (μM) are given for NS3 protease inhibition.

Table 2. Inhibition of NS3 Protease

		cmpd #	K_i
10			40
		1b1	21
15		1b2	640
		1c	0.5

K_i values (nM) are given for NS3 protease inhibition

As shown in the tables, the difluoro analogs 1a (K_i 1.5 μ M) and 1b¹ (K_i 21nM) had essentially equal affinity to their cysteine counterparts (K_i 0.7 μ M and 40nM respectively). 1b², the P₁ diastereomer of 1b¹ had reduced affinity.

The aldehyde 1c had K_i 0.5nM, and was a reversible competitive inhibitor with a slow K_{off} (1.4x10⁻³ S⁻¹) typical of covalent inhibitors of serine protease. Table 3 shows a comparison of the kinetic parameters of the acid 1b¹ with the aldehyde analogue 1c. The relatively long half-life of the protease-aldehyde complex (8 mins) compared to the (presumably) non-covalent protease-acid complex (<10s) is consistent with the formation of a

covalent bond between the active site serine and the aldehyde carbonyl group.

Table 3. Comparison of kinetic parameters for a non-covalent (**1b**) and covalent (**1c**) inhibitors of NS3 protease^a

cmpd	X	K _i (nM)	k _{on} (M ⁻¹ s ⁻¹)	k _{off} (s ⁻¹)	t _{1/2}
1b ¹	OH	21	2.2 × 10 ⁶	nd	< 10 s
1c	H	0.5	2.6 × 10 ⁶	1.4 × 10 ⁻³	8 min

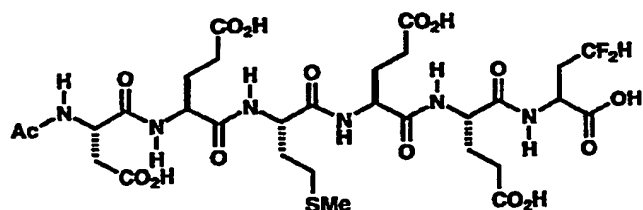
^aK_i values, k_{on} (on-rate), K_{off} (off-rate), and t_{1/2} (half-life of complex) are given for NS3 protease inhibition.

IC₅₀ values were determined for all of compounds 1a to 1m and are set out overleaf in Table 4.

TABLE 4

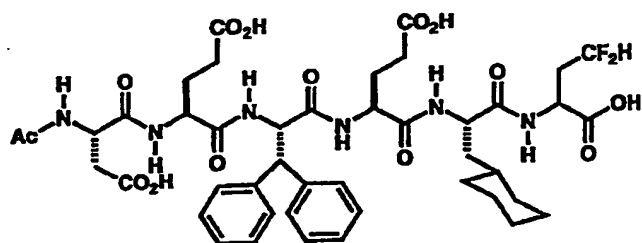
Compound

IC₅₀



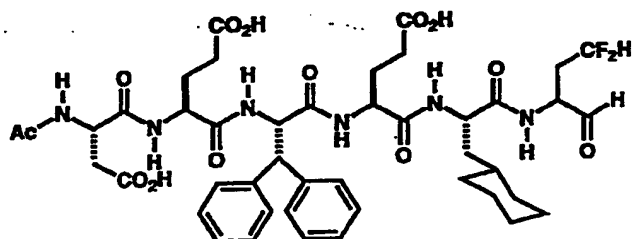
1a

3 μM



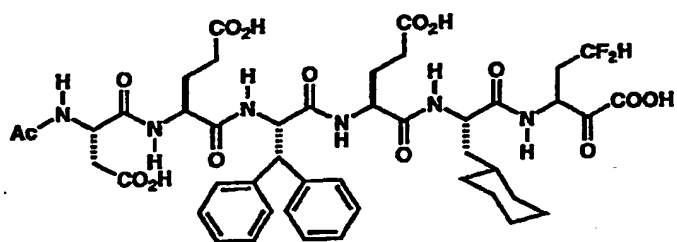
1b¹ (L)
1b² (D)

20 nM
1 μM



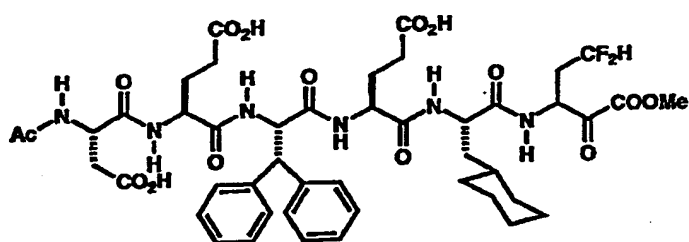
1c (L)
(D)

0.5 nM
43 nM



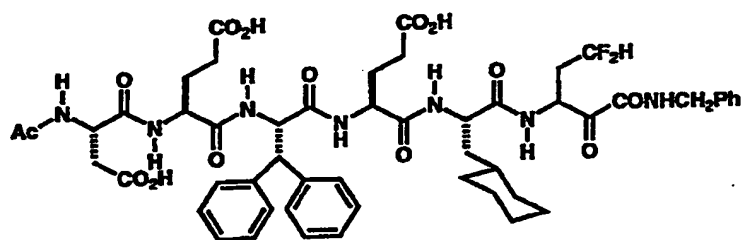
1d

0.4 nM



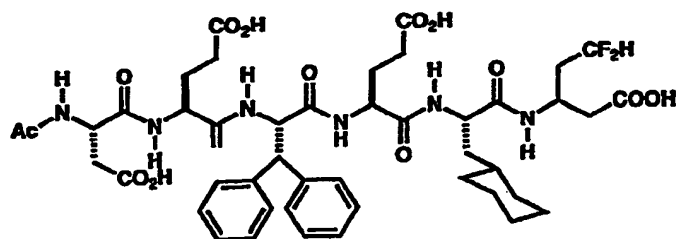
1j

800 nM



1e

9 nM

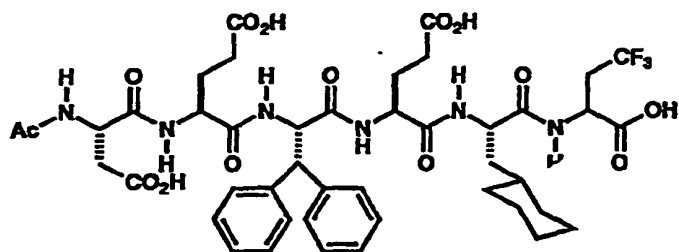


1k

3 μM

TABLE 4 continued

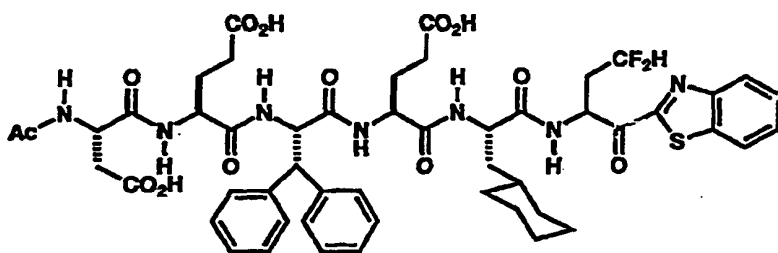
5



1f

110 nM

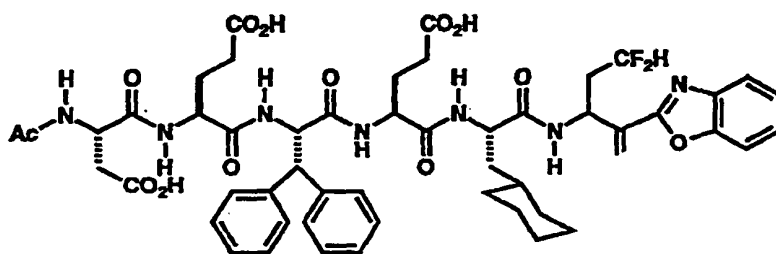
10



1g

125 nM

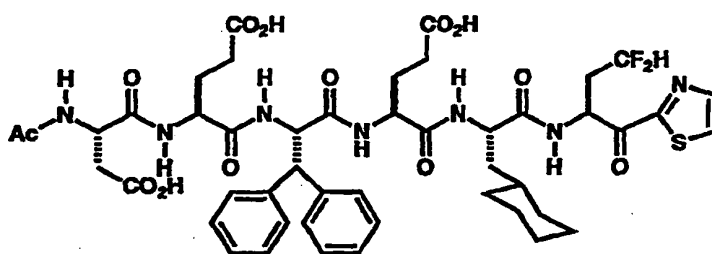
15



1h

600 nM

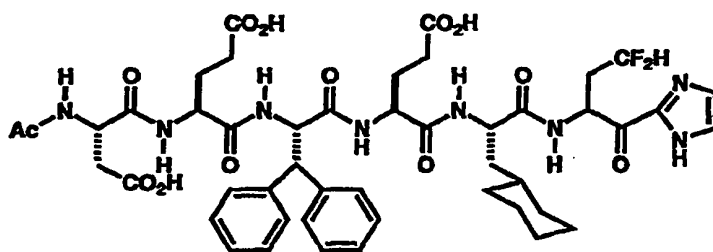
20



1i

1 μM

25



1j

6 μM

30

Notes to table:

Compounds 1d, 1j, 1e and 1k were produced as

diastereomeric mixtures which were subsequently separated

35

5 by chromatography. The IC_{50} value given is for the more active diastereomer which is presumed to have L stereochemistry at the P1 position.

10 The data for compound 1f are for the diastereomer having L stereochemistry at position 1.

Data for compounds 1g, 1l, 1h and 1m are for diastereomeric mixtures.

15

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25

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